

SAMPLE PAPER - 2

Time : 3½ Hrs.

MM : 100

GENERAL INSTRUCTIONS

- I. Question paper comprises of three sections, A, B and C you are to attempt all the three sections.
- II. All questions are compulsory.
- III. All questions of section A, section B and section C are to be attempted separately.
- IV. Question numbers 1 to 3 in section A are one mark each, to be answered in one word or one sentence.
- V. Question numbers 4 to 7 are two marks each, to be answered in about 30 words.
- VI. Question numbers 8 to 19 are three marks each, to be answered in about 50 words.
- VII. Question numbers 20 to 24 are five marks each, to be answered in about 70 words.
- VIII. Question numbers 25 to 42 in section B are MCQs based on practical skills. Each question is a one mark question.
- IX. Question numbers 43 and 44 in section C are OTBA based questions. Each question is of five marks. The open text material for these questions has been provided on pages (A-19-A-24)

SECTION - A

1. What conclusion would you derive from the observation, that in Rutherford's scattering experiment, most of the alpha-particles passed through gold foil without any deflection?
2. Name the organism in which asexual reproduction takes place by budding.
3. Define power.
4. What are the advantages of classifying organisms?
5. Acute diseases are severe but are less harmful than chronic diseases. Why?
6. Why a sound cannot be heard on the moon?
7. How is thrust different from pressure?
8. The relative atomic mass of boron is 10.8. Calculate the percentage of its isotopes $^{10}_5B$ and $^{11}_5B$ occurring in nature.
9. How many moles of Al atoms are needed to combine with 1.58 mole of O atoms to make aluminium oxide, Al_2O_3 ?
10. Copper oxide was prepared by the following methods:
 - (a) In one case, 1.75 g of the metal were dissolved in nitric acid and igniting the residual copper nitrate yielded 2.19 g of copper oxide.
 - (b) In the second case, 1.14 g of metal dissolved in nitric acid were precipitated as copper hydroxide by adding caustic alkali solution. The precipitated copper hydroxide after washing, drying and heating yielded 1.43g of copper oxide.
 - (c) In the third case, 1.45 g of copper when strongly heated in a current of air yielded 1.83 g of copper oxide.
Show that the given data illustrate the law of constant composition.
11. Why are the protozoan placed in kingdom-Protista and not in kingdom-Animalia?
12. Classify the following organisms based on the absence/presence of true coelom (i.e., acoelomate, pseudocoelomate and coelomate) *Spongilla*, *Sea anemone*, *Planaria*, *Liver fluke* *Wuchereria*, *Ascaris*, *Nereis*, earthworm, Scorpion, Birds, Fishes, horse.
13. What is antibiotic Penicillin? Give its function.
14. List any two differences between transverse and longitudinal waves. What is the nature of sound waves in air?

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Wenbin Ji

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