

109. When a charged particle in motion enters in a uniform magnetic field perpendicularly then its (1) Speed changes (2) Velocity changes (3) K.E. changes (4) Acceleration does not change	117. Which of the following salt does not contain the water of crystallization? (1) Blue Vitriol (2) Baking soda (3) Washing soda (4) Gypsum
110. The frequency of seconds pendulum is (1) 0.5 Hz (2) 1.0 Hz (3) 2.0 Hz (4) 1.5 Hz	118. The ion of an element has 3 positive charge, 27 mass-number and 14 neutrons. What is the number of electrons in this ion? (1) 13 (2) 14 (3) 10 (4) 16
111. The structure of solids is investigated by using (1) Cosmic rays (2) X-rays (3) Gamma rays (4) Infrared rays	119. Which one of the following is the natural fruit ripening hormone? (1) Ethane (2) Ethene (3) Ethyne (4) Carbide
112. Two bodies with kinetic energy in the ratio of 4 : 1 are moving with equal linear momentum. The ratio of their masses is (1) 1 : 2 (2) 1 : 1 (3) 4 : 1 (4) 1 : 4	120. Which of the following contains acidic hydrogen? (1) Ethene (2) Ethane (3) Ethyne (4) But-2-yne
113. A ray of light is incident on the surface of separation of a medium with the velocity of light in air at an angle $45^\circ$ and is refracted in the medium at an angle $30^\circ$ . What will be the velocity of light in the medium. (1) $1.96 \times 10^8$ m/s (2) $2.12 \times 10^8$ m/s (3) $3.18 \times 10^8$ m/s (4) $3.33 \times 10^8$ m/s	121. Silver jewellery becomes black on prolonged exposure to air. It is due to the formation of (1) $\text{Ag}_2\text{N}$ (2) $\text{Ag}_2\text{O}$ (3) $\text{Ag}_2\text{S}$ and $\text{Ag}_2\text{N}$ (4) $\text{Ag}_2\text{S}$
114. Analyse the given statements and choose the correct option. <b>Statement-I</b> : When current is represented by a straight line, the magnetic field will be circular. <b>Statement-II</b> : According to Fleming's left hand rule, the direction of the force is parallel to the magnetic field. (1) Both statement-I and statement-II are correct and statement-II is the correct explanation of statement-I. (2) Both statement-I and statement-II are true but statement-II is not the correct explanation of statement-I. (3) Statement-I is true but statement-II is false. (4) Statement-I is false but statement-II is true.	122. What is the mass of oxygen required to react completely with 15 g of $\text{H}_2$ gas to form water? (1) 140 g (2) 115 g (3) 107.5 g (4) 120 g
115. Which one of the following is the smallest in size? (1) $\text{N}^+$ (2) $\text{O}^-$ (3) $\text{F}^-$ (4) $\text{Na}^+$	123. Percentage purity of a sample of gold is 85%. How many atoms of gold are present in its 1 gram sample? (Atomic mass of gold = 197 u) (1) $2.6 \times 10^{21}$ (2) $2.6 \times 10^{22}$ (3) $3.0 \times 10^{21}$ (4) $4.5 \times 10^{22}$
116. Which chemical substance is added to LPG to help in detection of its leakage? (1) Isobutane (2) Ethanethiol (3) Propane (4) Hydrogen sulphide	124. The volume of 1 mole of an ideal gas at 25°C temperature and 1 bar pressure is: (1) 22.4 L (2) 22.7 L (3) 24.8 L (4) 24.4 L
	125. Which of the following solution can be stored in aluminium container? (1) $\text{MgSO}_4$ (aq) (2) $\text{ZnSO}_4$ (aq) (3) $\text{CuSO}_4$ (aq) (4) $\text{FeSO}_4$ (aq)
	126. What is the correct order of pH of aqueous solution of the following salts? (1) $\text{NaCl} = \text{Na}_2\text{CO}_3 = \text{NH}_4\text{Cl}$ (2) $\text{NaCl} < \text{Na}_2\text{CO}_3 < \text{NH}_4\text{Cl}$ (3) $\text{NH}_4\text{Cl} < \text{Na}_2\text{CO}_3 < \text{NaCl}$ (4) $\text{NH}_4\text{Cl} < \text{NaCl} < \text{Na}_2\text{CO}_3$

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