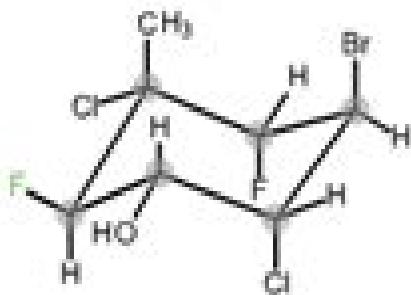


For the substituted cyclohexane compound given below, highlight the groups – by clicking on atoms – that are trans to the bromo substituent.



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 Explanation

For cyclohexane compounds drawn in a chair conformation, axial bonds are vertical, whereas equatorial bonds are parallel to two bonds within the ring. Thus, as drawn, the bromo substituent is in an axial position.

Keep in mind that trans groups are on opposite faces of the ring. Depending on the relative ring positions (i.e., 1,2 vs 1,3, etc.), the trans group might be in an axial or an equatorial position.

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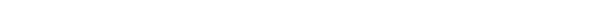
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