

Chemical Equations

- a shorthand way of reporting the details of a chemical reaction.

Reactants

- the starting substances in a chemical reactions
- placed on the left side of the equation

Products

- the substances starting during a chemical reaction
- placed on the right side of the equation

→

- arrow is read as yields or forms

Coefficients

- used to balance equations
- represent the number of molecules, F.U.'s, or atoms of the substance

Law of Conservation of Mass

Mass cannot be created or destroyed.



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