

# Introduction to Chemistry

## Section 1.1 The Stories of Two Chemicals

By 1940, the world had more than 1000 known elements.

Two acids and their stories follow just to give the examples the paragraph.

Acid	Properties	Source	Discovery Date
Hydrochloric acid	Strong acid	Minerals	1774
Hydrofluoric acid	Strongest acid	Minerals	1776

Hydrochloric acid \_\_\_\_\_ is usually a yellow liquid. This acid can dissolve almost any other metal to form a salt. This type is called the HCl \_\_\_\_\_. The acid type is called the HClO \_\_\_\_\_. This acid dissolves at proteins.

HCl \_\_\_\_\_ reacts \_\_\_\_\_.

Hydrofluoric acid (HF) \_\_\_\_\_ is usually a colorless liquid or the vapor part of the atmosphere. The name comes from a Latin word, *fluere*, which means "to flow". This acid reacts with most metals to form a salt. This acid is called the HF \_\_\_\_\_.

HF \_\_\_\_\_ is a colorless liquid. It can dissolve glass or metal and possibly other materials. The presence of this acid liquid called the HF \_\_\_\_\_ is dangerous because it will react with most materials.

By 1940, the world had more than 1000 known elements.

Two acids and their stories follow, beginning with sulfuric acid.

- 1. HCl is another name for hydrochloric acid.
- 2. HClO is also called hypochlorous acid. Bleach and chlorine.
- 3. HClO<sub>2</sub> is a hypochlorous acid salt.
- 4. HClO<sub>3</sub> is another name for chloric acid.
- 5. HClO<sub>4</sub> is another name for perchloric acid. Very acidic and dangerous.

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