

Name: _____

MULTIPLE CHOICE. Choose the one alternative that best completes the statement or answers the question.

- 1) A chemical reaction that absorbs heat from the surroundings is said to be _____ and has a _____ ΔH at constant pressure.
- endothermic, positive
 - endothermic, negative
 - exothermic, negative
 - exothermic, positive
 - exothermic, neutral

- 2) The reaction



is _____, and therefore heat is _____ by the reaction.

- exothermic, released
 - exothermic, absorbed
 - endothermic, released
 - endothermic, absorbed
 - thermoneutral, neither released nor absorbed
- 3) The value of ΔH° for the reaction below is -126 kJ . How much heat (in kJ) is released when 2.00 mol of NaOH is formed in the reaction?



- 252
 - 126
 - 7.8
 - 63
 - 3.9
- 4) The value of ΔH° for the reaction below is -790 kJ . The enthalpy change accompanying the reaction of 0.95 g of S is _____ kJ.



- 23
 - 12
 - 12
 - 23
 - 790
- 5) The value of ΔH° for the reaction below is -6535 kJ . How many kJ of heat are released in the combustion of 16.0 g of C_6H_6 (l)?



- 673
- 2.68×10^3
- 5.23×10^4
- 6535
- 1.34×10^3

Pearson Chemical Reaction Test Ch 6

Anthony S. Fauci



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1. Understanding the eBook Pearson Chemical Reaction Test Ch 6
 - The Rise of Digital Reading Pearson Chemical Reaction Test Ch 6
 - Advantages of eBooks Over Traditional Books
2. Identifying Pearson Chemical Reaction Test Ch 6
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Pearson Chemical Reaction Test Ch 6
 - User-Friendly Interface
4. Exploring eBook Recommendations from Pearson Chemical Reaction Test Ch 6
 - Personalized Recommendations
 - Pearson Chemical Reaction Test Ch 6 User Reviews and Ratings
 - Pearson Chemical Reaction Test Ch 6 and Bestseller Lists
5. Accessing Pearson Chemical Reaction Test Ch 6 Free and Paid eBooks

- Pearson Chemical Reaction Test Ch 6 Public Domain eBooks
 - Pearson Chemical Reaction Test Ch 6 eBook Subscription Services
 - Pearson Chemical Reaction Test Ch 6 Budget-Friendly Options
6. Navigating Pearson Chemical Reaction Test Ch 6 eBook Formats
- ePub, PDF, MOBI, and More
 - Pearson Chemical Reaction Test Ch 6 Compatibility with Devices
 - Pearson Chemical Reaction Test Ch 6 Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Pearson Chemical Reaction Test Ch 6
 - Highlighting and Note-Taking Pearson Chemical Reaction Test Ch 6
 - Interactive Elements Pearson Chemical Reaction Test Ch 6
8. Staying Engaged with Pearson Chemical Reaction Test Ch 6
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Pearson Chemical Reaction Test Ch 6
9. Balancing eBooks and Physical Books Pearson Chemical Reaction Test Ch 6
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Pearson Chemical Reaction Test Ch 6
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Pearson Chemical Reaction Test Ch 6
- Setting Reading Goals Pearson Chemical Reaction Test Ch 6
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Pearson Chemical Reaction Test Ch 6
- Fact-Checking eBook Content of Pearson Chemical Reaction Test Ch 6
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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