

11 Chemical Reactions

Big idea REACTIONS

11.1 Describing Chemical Reactions



For students using the Foundation edition, assign problems 1–17.

Essential Understanding Chemical reactions are represented by balanced chemical equations.

Lesson Summary

Introduction to Chemical Equations A chemical equation uses symbols, and sometimes words, to show the reactants and products of a chemical reaction.

- A skeleton equation uses chemical formulas to represent reactants and products, but it does not indicate the relative amounts of each.
- Anything that enters into a reaction, such as heat or a catalyst, but is not a reactant or product is shown above or below the yields arrow in the equation.

Balancing Chemical Equations Chemical equations are balanced to show that mass is conserved during chemical reactions.

- A balanced equation shows the relative amounts of reactants and products, and it contains equal numbers of each type of atom on both sides of the equation.
- Chemical equations are balanced by using coefficients in front of the chemical formulas for the reactants and the products in a skeleton equation.



BUILD Math Skills

Balancing Equations All chemical equations must be balanced because of the law of conservation of mass, which states that matter cannot be created or destroyed. So, the number of atoms that you start with at the beginning of the reaction must equal the number of atoms that you end up with.

For example, the reaction $2\text{Mg} + \text{O}_2 \rightarrow 2\text{MgO}$ follows the law of conservation of mass because you start with 2 magnesium atoms and 2 oxygen atoms and you end up with 2 magnesium atoms and 2 oxygen atoms. You can think of the \rightarrow as an = sign.



Turn the page to learn more about balancing equations.

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and hemostasis This new edition details the parts and functions of the cell shows how to accurately identify cells covers

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and thrombosis concepts and disorders Easy to follow and understand this book also covers key topics including working in

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diagnostics and laboratory testing of blood cells and body fluid cells Content throughout the text reflects the latest

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Major revision of the Automated Blood Cell Analysis chapter

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 - The Rise of Digital Reading Pearson Chemistry Reading And Study Workbook Answers Chapter 14
 - Advantages of eBooks Over Traditional Books
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 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Pearson Chemistry Reading And Study Workbook Answers Chapter 14
 - User-Friendly Interface
4. Exploring eBook Recommendations from Pearson Chemistry Reading And Study Workbook Answers Chapter 14
 - Personalized Recommendations
 - Pearson Chemistry Reading And Study Workbook Answers Chapter 14 User Reviews and Ratings
 - Pearson Chemistry Reading And Study Workbook Answers Chapter 14 and Bestseller Lists
5. Accessing Pearson Chemistry Reading And Study Workbook Answers Chapter 14 Free and Paid eBooks
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 - Pearson Chemistry Reading And Study Workbook Answers Chapter 14 eBook Subscription Services
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 - Highlighting and Note-Taking Pearson Chemistry Reading And Study Workbook Answers Chapter 14
 - Interactive Elements Pearson Chemistry Reading And Study Workbook Answers Chapter 14
8. Staying Engaged with Pearson Chemistry Reading And Study Workbook Answers Chapter 14
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Pearson Chemistry Reading And Study Workbook Answers Chapter 14
9. Balancing eBooks and Physical Books Pearson Chemistry Reading And Study Workbook Answers Chapter 14
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Pearson Chemistry Reading And Study Workbook Answers Chapter 14
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 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
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 - Setting Reading Goals Pearson Chemistry Reading And Study Workbook Answers Chapter 14
 - Carving Out Dedicated Reading Time
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 - Distinguishing Credible Sources
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