

Chapter 10 test**Matching**

Match each item with the correct statement below.

- | | |
|----------------------------|--------------------------------------|
| a. representative particle | d. percent composition |
| b. mole | e. standard temperature and pressure |
| c. Avogadro's number | f. empirical formula |

- _____ 1. the number of representative particles of a substance present in 1 mole of that substance
_____ 2. an atom, an ion, or a molecule, depending upon the way a substance commonly exists
_____ 3. the SI unit used to measure amount of substance
_____ 4. 0°C and 1 atm
_____ 5. the percent by mass of each element in a compound
_____ 6. the smallest whole number ratio of the atoms in a compound

Multiple Choice

Identify the letter of the choice that best completes the statement or answers the question.

- _____ 7. What SI unit is used to measure the number of representative particles in a substance?
a. kilogram
b. ampere
c. kelvin
d. mole
- _____ 8. How many hydrogen atoms are in 5 molecules of isopropyl alcohol, C_3H_7O ?
a. $5 \times (6.02 \times 10^{23})$
b. 5
c. 35
d. $35 \times (6.02 \times 10^{23})$
- _____ 9. All of the following are equal to Avogadro's number EXCEPT _____.
a. the number of atoms of bromine in 1 mol Br_2
b. the number of atoms of gold in 1 mol Au
c. the number of molecules of nitrogen in 1 mol N_2
d. the number of molecules of carbon monoxide in 1 mol CO
- _____ 10. How many moles of tungsten atoms are in 4.8×10^{25} atoms of tungsten?
a. 8.0×10^{-2} moles
b. 8.0×10^{-1} moles
c. 1.3×10^{-4} moles
d. 1.3×10^{-2} moles
- _____ 11. How many atoms are in 0.075 mol of titanium?
a. 1.2×10^{23}
b. 2.2×10^{24}
c. 6.4×10^2
d. 4.5×10^{22}
- _____ 12. How many molecules are in 2.10 mol CO_2 ?
a. 2.53×10^{24} molecules
b. 3.79×10^{23} molecules
c. 3.49×10^{-24} molecules
d. 1.26×10^{24} molecules

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**Mary Ruth Brown, Eugenie Lair
Moss, Karin Drudge Bright**



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Table of Contents Pearson Chemistry Textbook Chapter 10 Test Answers

1. Understanding the eBook Pearson Chemistry Textbook Chapter 10 Test Answers
 - The Rise of Digital Reading Pearson Chemistry Textbook Chapter 10 Test Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Pearson Chemistry Textbook Chapter 10 Test Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Pearson Chemistry Textbook Chapter 10 Test Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Pearson Chemistry Textbook Chapter 10 Test Answers
 - Personalized Recommendations
 - Pearson Chemistry Textbook Chapter 10 Test Answers User Reviews and Ratings

- Pearson Chemistry Textbook Chapter 10 Test Answers and Bestseller Lists
- 5. Accessing Pearson Chemistry Textbook Chapter 10 Test Answers Free and Paid eBooks
 - Pearson Chemistry Textbook Chapter 10 Test Answers Public Domain eBooks
 - Pearson Chemistry Textbook Chapter 10 Test Answers eBook Subscription Services
 - Pearson Chemistry Textbook Chapter 10 Test Answers Budget-Friendly Options
- 6. Navigating Pearson Chemistry Textbook Chapter 10 Test Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Pearson Chemistry Textbook Chapter 10 Test Answers Compatibility with Devices
 - Pearson Chemistry Textbook Chapter 10 Test Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Pearson Chemistry Textbook Chapter 10 Test Answers
 - Highlighting and Note-Taking Pearson Chemistry Textbook Chapter 10 Test Answers
 - Interactive Elements Pearson Chemistry Textbook Chapter 10 Test Answers
- 8. Staying Engaged with Pearson Chemistry Textbook Chapter 10 Test Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Pearson Chemistry Textbook Chapter 10 Test Answers
- 9. Balancing eBooks and Physical Books Pearson Chemistry Textbook Chapter 10 Test Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Pearson Chemistry Textbook Chapter 10 Test Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Pearson Chemistry Textbook Chapter 10 Test Answers
 - Setting Reading Goals Pearson Chemistry Textbook Chapter 10 Test Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Pearson Chemistry Textbook Chapter 10 Test Answers
 - Fact-Checking eBook Content of Pearson Chemistry Textbook Chapter 10 Test Answers
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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