

NUCLEAR CHEMISTRY WORKSHEET

Part A: Completing the Nuclear Decay Reaction

For each of the atoms listed below, complete the decay reaction by solving for ΔZ or other missing information. Remember that the mass and protons on each side of the arrow need to equal each other.



Part b: Writing Nuclear Decay Reactions

Write equations for the following nuclear decay reactions. Make sure that both mass numbers and atomic numbers are balanced on each side

11. Decay of polonium-218 by alpha (α) emission
12. Decay of carbon-14 by beta (β emission)
13. The alpha decay of radon-198
14. The beta (β) decay of uranium-237
15. The alpha decay of plutonium- 243
16. Potassium-40 can decay to form calcium-40. Write the balanced equation for this nuclear reaction.
17. Write a balanced equation for the nuclear reaction in which neon-23 decays to form sodium-23.
18. Write a balanced nuclear equation for the radioactive decay of iron-60 to give cobalt-60.
19. Write a nuclear equation for the following radioactive process; beta emission by argon-37
20. When uranium-234 decays by emitting an alpha particle, it becomes?

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