



Problem Set 4
Chapters 4 and 25

Due Thursday Oct 24th

Nuclear Chemistry
Essay Due Oct 30th

Test Wednesday October 30th

- Your work is to be done on separate sheets of paper.
- All of the papers should be stapled together in the upper left hand corner with this question sheet at the back of the packet.
- Each of your responses should be numbered the same way as it appears on this packet.

Conceptual Questions

- 1) State the law of conservation of mass.
- 2) Why are atoms unstable?
- 3) How do unstable atoms gain stability?
- 4) A radioactive element having atomic number 82 and atomic mass 214 loses a beta particle, $0_{-1}\beta$. What is the resulting particle?
- 5) What type of decay is the reaction below, and what isotope completes this nuclear equation?
 $27_{13}\text{Al} + 2_1\text{H} \rightarrow 4_2\text{He} + ?$

Practice Questions

- 6) Complete this chart on the three types of radioactive decay

Type	Symbol	Mass (amu)	Charge
Alpha			
Beta			
Gamma			

- 7) If the radioactive atom 238_{92}U emits an alpha particle, the atom remaining is represented by
(A) 238_{92}U (B) 236_{92}Th (C) 234_{92}U (D) 234_{90}Th (E) 234_{91}Pa
- 8) What term completes this equation? $30_{14}\text{Si} + ? \rightarrow 32_{16}\text{S} + 0_{-1}e$
(A) 4_2He (B) 2_1H (C) 1_1H (D) $0_{-1}e$
- 9) What particle is emitted when lead-214, 214_{82}Pb , undergoes natural radioactive decay to bismuth-214, 214_{83}Bi ?
(A) an alpha particle, 4_2He (C) a proton, 1_1H
(B) a beta particle, $0_{-1}e$ (D) a neutron, 1_0n

Extension Question

- 10) Recently Alexander Valterovich Litvinenko was poisoned by radioactive polonium-210.
 - a. Research who he was and why he was poisoned.
 - b. Find out what polonium-210 does to people.

Nuclear Chemistry Worksheet Chapter 25

G Orfield



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Spreadsheet Chemistry O. Jerry Parker, Gary L. Breneman, 1991 **Holt Chemistry** Ralph Thomas Myers, 2004

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