

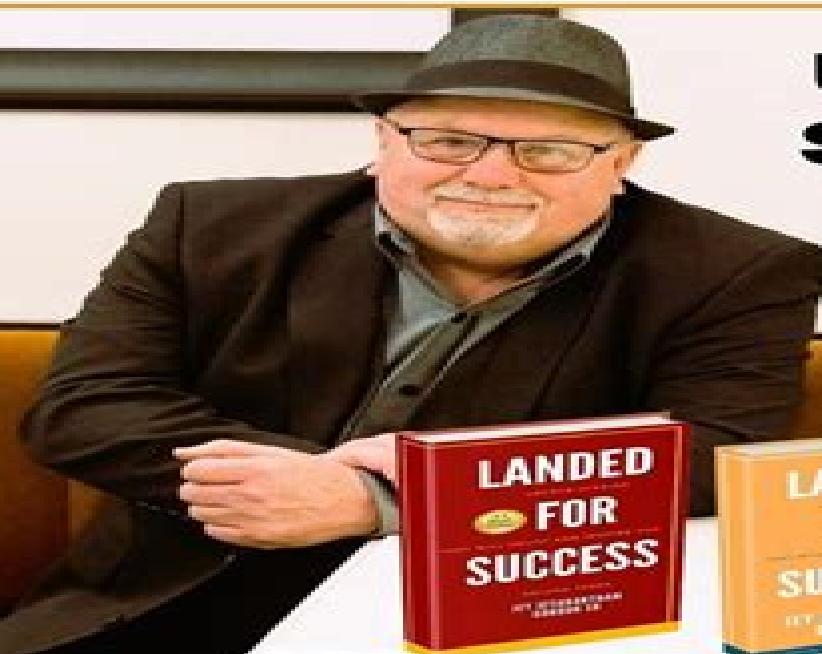
Landed for Success

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Tom Chessler recently received an Honorary Doctorate Degree in International Business and is a contributing author in two Landed For Success International Best-Selling Books. Dr. Tom Chessler, Rise Up Media & Marketing provides "Online News Sources & Publishers" with Media-Worthy stories and content.

Personal Finance Success International Bestseller

Peipei Pang

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web $m \times c \times \Delta t$ equation 1 where m is the total mass of the solution solute plus solvent c is the specific heat of the solution and Δt is the observed temperature change the specific heat of the solution is generally assumed to be the same as that of

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web the heat of reaction Δh_{soln} is written after the products in units of kj mol rxn for example sodium hydroxide dissolves

exothermically releasing 44.2 kilojoules per mole dissolved this process is represented as $\text{NaOH}(s) \rightarrow \text{NaOH}(aq)$ $\Delta H_{\text{soln}} = 44.2 \text{ kJ/mol rxn}$ write equations to similarly represent the dissolving process for

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